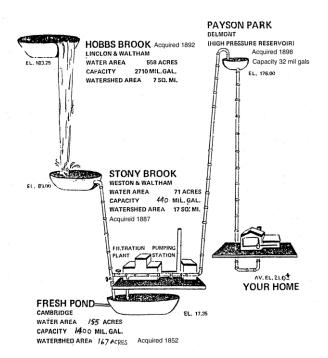
# City of Cambridge, Water Department

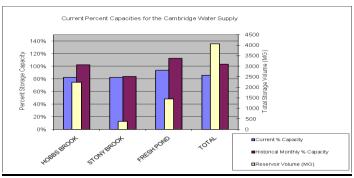
250 Fresh Pond Parkway, Cambridge, MA 02138
Monthly Water Quantity and Quality Report
November 2009



The data contained in this report is for informational purposes only subject to verification and not intended for regulatory compliance.

SOURCE of CAMBRIDGE WATER SUPPLY





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David Kaplan Watershed Technician Watershed Division dkaplan@cambridgema.gov

### **Water Supply**

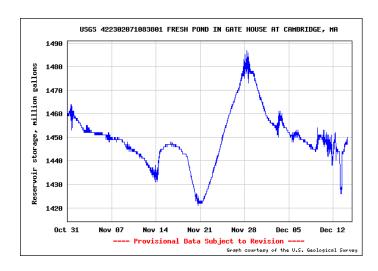
The Cambridge water supply system consists of four reservoirs and their tributaries located in Lexington, Waltham, Lincoln, Weston, Cambridge and Belmont. The two primary water sources, Hobbs Brook Reservoir and Stony Brook Reservoir, flow to the terminal reservoir, Fresh Pond, located in Cambridge, via the Stony Brook conduit. The water is then purified and pumped to Payson Park Reservoir, two 16 million gallon drinking water clearwells located in Belmont at a maximum elevation of 181 feet, where it is further disinfected with chloramines and distributed to the city by gravity. The largest of the reservoirs, Hobbs Brook, reaches its maximum elevation at 181.3 feet above sea level, its maximum depth at approximately 25 feet, and at full capacity, holds approximately 2.5 billion gallons of water. Stony Brook Reservoir reaches its maximum elevation at 80.6 feet above sea level, its deepest point is at approximately 35 feet, and at full capacity, it contains roughly 455 million gallons of waterrevised 2009. Fresh Pond Reservoir reaches its maximum elevation at 17 feet above sea level, its maximum depth at 50 feet, and at full capacity, holds roughly 1.5 billion gallons.

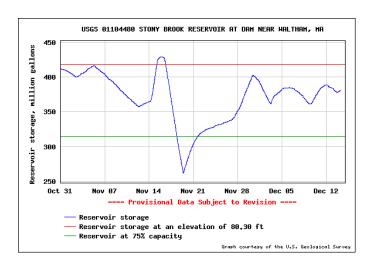
While the Watershed's primary storage reservoir is Hobbs Brook, in the winter and spring months it is largely unused. Due to its vast size, and relatively small watershed, Hobbs Brook is slow to fill up. This winter hiatus is necessary for Hobbs to regain the water it uses in the summer months. Conversely, Stony Brook is relatively small compared to its large watershed and fills much faster than Hobbs Brook. Due to this condition, Stony Brook is used in place of Hobbs during the winter months. During times of high water flow, the Cambridge Watershed, via Stony Brook, overflows its surplus water into the Charles River.

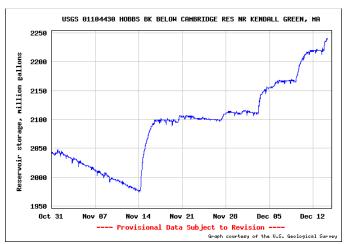
Current Conditions: 12/14/09

The reservoir system is at 86% capacity. The reservoir system is currently at 103% of historical capacity for this time of year. Hobbs Brook Reservoir current storage volume is 2,240 Million Gallons (MG) or 82% and filling, Stony Brook Reservoir has a current storage volume is 382 MG or 82% and Fresh Pond Reservoirs current storage volume is 1,448 MG or 93%. The average daily demand is 11.7 Million Gallons per Day (MGD). Days of supply remaining without recharge: 260 or 8.7 months.

# Reservoir storage levels



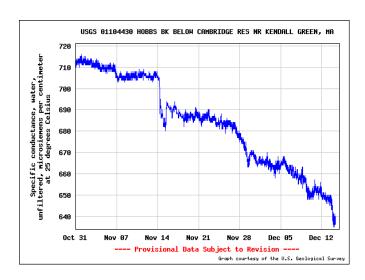




# **Reservoir Water Specific Conductance**







### **Water Demand**

-Cambridge City Population: 101,355

Million Gallons (MG) - Gallons per Minute (GPM)

Maximum instantaneous demand:

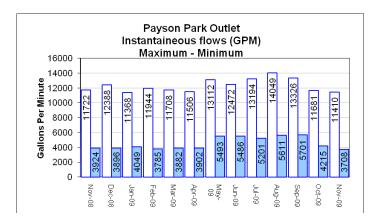
11/10/09 at 07:45 am: 11,410 GPM

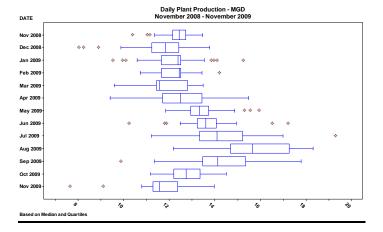
Minimum instantaneous demand:

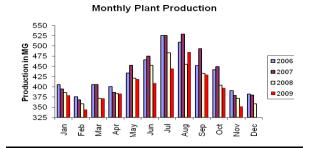
11/23/09 at 03:08 am: 3,708 GPM

Average daily water demand for 2009 (to date): 13.20 MG Average daily water demand 2008: 13.33 MG Average daily water demand November 2009: 11.71 MG 12.40 MG Average daily water demand November 2008: Monthly Water Demand November 2009: 351 MG Monthly Water Demand November 2008: 371MG Water produced for 2009 (to date): 4,407 MG Total water produced for 2008: 4,878 MG

### **Daily Payson Reservoir Outlet Flow:**







## **Source Water Quality**

### Hobbs Brook Reservoir at intake

Average: 2 CFU/100mL E-Coli Bacteria-TOC Average: 3.4 mg/L UV 254 Average: 0.111 A/cm Aluminum: Average: 0.039 mg/L Sodium: Average: 120 mg/L Average: 230 mg/L Chloride Average: 0.106 mg/L Bromide: Average: 0.080 NTU Turbidity-Conductivity-Average: 679 umhos/cm

pH-Average: 6.95

### Stony Brook Reservoir at intake

E-Coli Bacteria-Average: 37 CFU/100mL TOC Average: 5.47 mg/L UV 254 Average: 0.243 A/cm Aluminum Average: 0.061mg/L Sodium Average: 83 mg/L Chloride Average: 166 mg/L Average: 0.078 mg/L **Bromide** Turbidity Average: 1.6 NTU Conductivity-Average: 473 umhos/cm

Average: 6.80 pH-

#### Fresh Pond Reservoir at intake

E-Coli Bacteria-Average: 15 CFU/100mL TOC Average: 3.48 mg/L UV254 Average: 0.115 A/cm SUVA Average: 3.30 Average: 72.6 mg/L Sodium Chloride Average: 146 mg/L Average: 0.078 mg/L **Bromide** Alkalinity (as CaCO3) Average: 30.5 mg/L Aluminum Average: 0.015 mg/L Turbidity-Average: 0.411 NTU Conductivity-Average: 512umhos/cm pH-Average: 7.15

### Cryptosporidium and Giardia

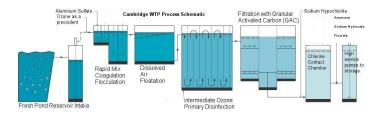
Average: 10°C

Temperature:

As part of the Long Term 2 Enhanced Surface Water Treatment Rule monitoring Cambridge has collected 24 samples for Cryptosporidium and Giardia since February 2006. The monitoring is population based. Hence systems with populations >100,000 must collect 24 raw water samples over a two year period. The Cryptosporidium results are the basis for bin assignment on the Running Annual Average (RAA). Bin Classification: Systems with an RAA Crypto concentrations of <0.75 (Oo) cysts per liter are placed in Bin 1 for which no additional treatment is required. To date none of the samples have detected any Cryptosporidium and one sample out of 24 detected Giardia at 0.1 cysts/L.

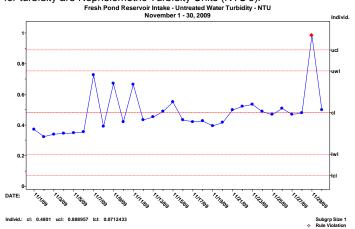
### Endocrine Disruptors, Pharmaceuticals, Personal Care Products

In response to the concern about the possibility of these unregulated compounds in municipal tap water, the Water Department tested the tap water in March 2008 for 86 compounds and did not detect any in the tap water. Follow up sampling in September 2008 detected the following compounds in the tap water: Acetaminophen at 0.019 micrograms per liter (ug/L) and Nicotine at 0.007 ug/L, Monitoring will continue twice a year. March 2009: Cotinine at 0.001 ug/L. September 2009: No Detects



### **Turbidity**

Turbidity is a measure of suspended and colloidal particles including clay, silt, and inorganic matter, algae, and microorganisms. Turbidity is determined by a technique involving the measurement of light scattered at right angles in a water sample. The more of the source light that is scattered the more (the higher) the turbidity. The units of measurement for turbidity are Nephelometric Turbidity Units (NTU's).



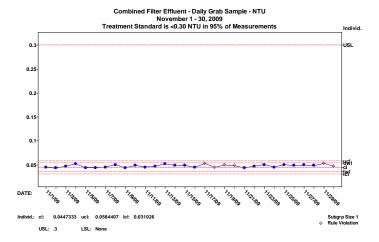
# Finished Water Quality

### **Turbidity**

The Surface Water Treatment Rule (SWTR) establishes performance goals for finished water turbidity. The pretreatment turbidity goal is to be less than 1 NTU. A conventional filtration plant is considered in compliance if the filtered water turbidity is less than 0.3 NTU in 95 % of its samples.

Turbidity is measured through the treatment process as a measure of treatment effectiveness. In the watershed and in the reservoirs turbidity may indicate the presence of silt from storm events or the presence of algae. In the pretreatment e.g. rapid mix, flocculation, and Dissolved Air Flotation (DAF)) portion of the plant turbidity is used indicator of process efficiency. Turbidity of the filter effluent is used both as a process efficiency and regulatory indicator of performance.

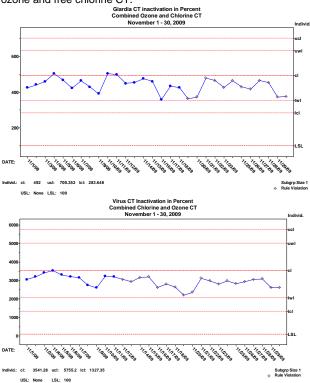
The effects of turbidity depend on the nature of the matter that causes the turbidity. High levels of particulate matter may have higher chlorine demand or may protect bacteria from the disinfectant effects of ozone and chlorine, thereby interfering with the disinfectant residual throughout the distribution system. The turbidity through the cycle of each filter run is an indicator of the overall effectiveness of the filter process.



### **Primary Disinfection - Ozone**

The CWD provides disinfection to achieve the EPA requirement for 99.9% inactivation of Giardia cysts and 99.99% inactivation of viruses in drinking water. Instead of measuring or counting Giardia and viruses, compliance is determined by a system operational standard, the measurement of the disinfection process. EPA has established a set of criteria for each disinfectant (ozone, free chlorine, and chloramines).

They are stated as CT values where C is concentration and T is time. The concentration C of the disinfectant in the water over time T yields a measure of the effectiveness of disinfection, CT. The required CT varies with the disinfectant type, water temperature, pH, and other factors. CWD measures CT in three places, intermediate ozone, free residual chlorine in the clearwell, and chloramines through the Payson Park Reservoir. The goal is to meet the minimum CT requirements with the intermediate ozone system at a concentration of 1.5 mg/L (milligram per liter) Ozone. The CT credited from the other two sources provides redundancy to the system. The following two graphs show the combined ozone and free chlorine CT.



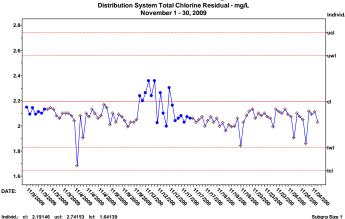
### **Secondary Disinfection - Chloramines**

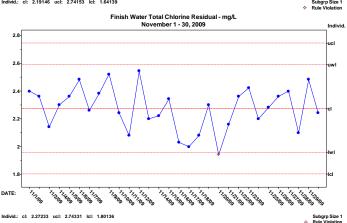
Regulations require a minimum of a 0.2 mg/L concentration of disinfectant throughout the distribution system. The higher values reflect the operational need for disinfection with free chlorine after the biological filters. A 15% solution of Sodium Hypochlorite is added at a concentration of 3.5 mg/L at the entrance the clearwell. The typical chlorine demand is approximately 1 mg/L this leaves a free residual chlorine concentration of 2.5 mg/L available for disinfection in the clearwell. It is this concentration of chlorine that's mixed with ammonia to create the level of chloramines measured as total residual chlorine using the HACH DPD Method.

### Chloramination

The practice of adding ammonia to chlorinated water is called Chloramination. This process is recognized for taste and odor control to reduce the undesirable medicinal taste of chlorinated water. It was first used in Greenville, Tennessee in 1926. This process can contribute to taste and odor control problems if not properly controlled. The formation of di- and trichloramines species is minimized by controlling the chlorine and ammonia ratios (3 to 4:1). A 30 % solution of Ammonium Hydroxide is added at a concentration of 0.5 mg/L. CWD's target chlorine to ammonia ratio is 4.5:1

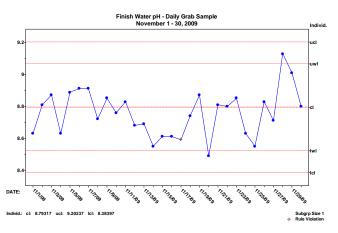
### Distribution Chlorine Residual - mg/L

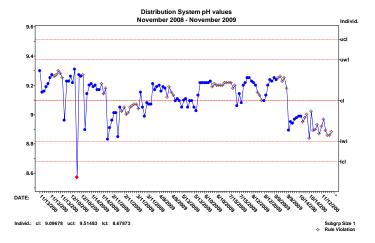




### Corrosion Control - pH adjustment

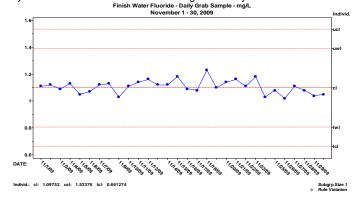
The National Primary Drinking Water Regulations (NPDWR) - Lead and Copper Rule establishes limits to the amount of lead and copper that may be in drinking water at the consumers tap. The Action Level for Lead is 15 ug/L (micrograms per liter). CWD is in compliance with the 2008 round of reduced sampling. CWD's 90th percentile is 9 ug/L. The Action Level for Copper is 1300 ug/L. CWD's 90th percentile was 32ug/L. Cambridge meets the requirements by reducing corrosiveness of the water by adjusting the to pH 9 with a 50% solution of Sodium Hydroxide (as of 1/26/09) at a concentration of 22 mg/L. This combined with the natural occurring alkalinity, hardness and dissolved minerals in the water minimizes the leaching of lead and copper from service lines and home plumbing systems, the source of lead and copper at the consumer tap. The target for distribution system pH is 9.1. On 9/24/09 Finish Water pH Set point reduced to 8.8. The High Lift Pumps Seized due to Calcium Carbonate scaling. This scale forms when the water temperature is over 20°C. Lowering Caustic addition at this time reduces this scaling.





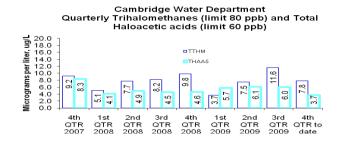
### **Fluoridation**

The Massachusetts Department of Health mandates that Drinking Water Systems fluoridate for the prevention of dental cavities. CWD adds a solution of 23% Hydrofluocylicic acid at a concentration of 1.0 mg/L. CWD targets the concentration at 1.1 mg/L. The Fluoride addition was stopped to determine the length of time the fluoride remains in the system. Estimated maximum water age is 12 days



### **Disinfection By-Products**

The incidence of waterborne diseases has been greatly reduced since the widespread implementation of drinking water disinfection. While a measurable public health benefit has been achieved, other potential risks may have been introduced. The presence of chloroform and other trihalomethanes (THMs) in finished drinking water was first associated with the chlorination of drinking water in 1974. It was discovered that in, addition to killing microorganisms disinfectants react with organic and inorganic substances naturally present in the water to produce a variety of disinfection by-products (DBPs), which include THMs. The DBPs associated with chlorination are THMs, haloacetic acids, haloacetonitriles and halopicrins. Ozonation may result in bromate formation.



Total Organ			-											
	CWD - TO	C Remova	l Performan	ce Summ	ary									
			DAF Train #1		DAF Train #2		Post- Ozone Train #1		Post- Ozone Train #2		In/Out	Pretreatment	Filter	
			Percent		Percent		Percent		Percent		Percent	Percent	Percent	
			removal		removal	Post Ozone	removal	Post Ozone	removal		Removal	Removal	Removal	
		<b>b</b> . DAF	efficiency	c. DAF	efficiency	d. Filter Influent	efficiency	e. Filter Influent	efficiency	f. Finished	Efficiency	Efficiency	Efficiency	Alum
Date	a. Raw	Train 1	1-b/a*100	Train 2	1-c/a*100	Train 1	1-d/b*100	Train 2	1-e/c*100	Water	1-f/a*100	((b+d)+(c+e))/2	(in/out -pretreat)	dose
	mg/L	mg/L	%	mg/L	%	mg/L	%	mg/L	%	mg/L	%	%	%	mg/L
11/13/2008	4.19	2.27	46	2.10	50	2.28	-0.6	2.11	-0.3	1.64	61	47	13.4	31
12/1/2008	4.29	2.41	44	2.14	50	2.69	-11.3	2.20	-3.1	1.73	60	40	19.9	31
12/8/2008	4.35	2.19	50	2.11	52	2.34	-6.8	2.25	-6.8	1.76	60	44	15.8	32
1/7/2009	4.11	2.04	50	2.02	51	2.06	-1.2	2.01	0.7	1.65	60	50	9.4	34
2/2/2009	4.14	2.09	50	2.03	51	2.02	3.2	2.07	-1.8	1.63	61	51	9.7	33
3/2/2009	3.97	2.27	43	2.11	47	2.03	10.4	2.05	2.8	1.75	56	51	4.6	25
4/7/2009	3.52	1.78	49	1.79	49	1.76	1.0	1.79	-0.2	1.45	59	50	9.0	26
5/7/2009	3.65	1.90	48	1.84	50	1.75	8.2	1.52	17.5	1.56	57	62	-4.3	26
6/1/2009	3.69	1.97	47	1.98	46	1.94	1.4	1.91	3.5	1.63	56	49	6.8	33
6/29/2009	3.54	1.82	48	1.88	47	1.70	7.0	1.73	8.0	1.49	58	55	2.7	30
7/31/2009	3.77	1.81	52	1.80	52	1.69	6.7	1.69	5.9	1.41	63	58	4.3	27
9/1/2009	3.80	1.81	52	2.01	47	1.56	13.8	1.50	25.4	1.36	64	69	-5.1	32
9/29/2009	3.91	2.09	47	1.99	49	1.91	8.6	1.81	9.0	1.61	59	57	2.2	29
10/19/2009	3.87	2.22	43	2.08	46	2.15	3.2	1.79	13.9	1.58	59	53	6.2	32
11/30/2009	4.03	1.96	51	1.83	55	1.92	2.0	1.82	0.55	1.51	63	54	8.3	30
Average	3.9	2.0	48.0	2.0	49.4	2.0	3.0	1.9	5.0	1.6	60	53	6.9	30.1
Jnits: Milligra	ms per Lite	er(mg/L)												
Percentages	s do not ac	d up to 10	00%											
OC samples	s a grab sa	mples taker	n accross the	plant at a	single time a	and not a single o	ohort of wa	ter						

# Cambridge Water Department Laboratory

Massachusetts Certification Number M-MA149

		Monthly	Monthly			Percent		
		Average	minimum	Samples	Total	Total		
		Chlorine	Chlorine	tested	samples	samples	Fecal	Public
		Residual	Residual	for	Coliform	Coliform	Coliform	Notification
		(mg/L)	(mg/L)	Coliform	positive	positive	positive	required
2008	October	2.17	1.62	100	0	0	0	no
	November	2.38	2.07	100	0	0	0	no
	December	2.30	1.84	100	0	0	0	no
2009	January	2.01	1.77	100	0	0	0	no
	February	2.09	1.94	100	0	0	0	no
	March	2.08	1.93	100	0	0	0	no
	April	2.06	1.89	100	0	0	0	no
	May	2.05	1.86	100	0	0	0	no
	June	2.21	1.93	100	0	0	0	no
	July	2.15	1.88	100	0	0	0	no
	August	2.02	1.48	100	0	0	0	no
	September	2.27	1.80	100	0	0	0	no
	October	2.30	1.78	100	0	0	0	no
	November	2.04	1.68	100	0	0	0	no

<u>Plant Discharge Permits</u>
Massachusetts Water Resource Authority (MWRA) Toxic Reduction And Control (TRAC) program regulates discharge to the sewer system. The TRAC program classifies CWD as a Significant Industrial User (SIU) and has three permits for discharge: 0101 Residuals, 0102 Laboratory, and 0103 Truck Wash.

The EPA regulates discharge of clarified backwash water to Fresh Pond under the National Pollution Discharge Elimination System (NPDES) Act

		Monthly Residuals -	Monthly Residuals -	Semi-annual Lab Waste -	Semi-annual Lab Waste -	Semi-annual Truck Wash -	Semi annual Truck Wash -	Quarterly
		0101	0101	0102	0102	0103	0103	NPDES
	Daily Maximum							3rd QTR
Parameter	Limit - <b>mg/L</b>	11/02/09	10/07/09	01/06/09	7/06/09	01/06/09	07/06/09	2009
Aluminum (NPDES)	Report Max. #	NR	NR	NR	NR	NR	NR	1.26
Cadmium	0.1	< 0.002	< 0.002	NR	NR	< 0.005	< 0.005	NR
Copper	1.5	0.27	0.17	< 0.010	< 0.010	0.022	0.028	NR
Chromium	1	0.055	0.050	NR	NR	<0.01	< 0.01	NR
Lead	0.2	0.025	0.005	< 0.01	10	0.027	0.013	NR
Nickel	1	0.064	0.041	< 0.025	< 0.025	< 0.025	< 0.025	NR
Silver	2	< 0.002	< 0.002	NR	NR	NR	NR	NR
Zinc	1	0.16	0.120	NR	NR	0.228	0.434	NR
Arsenic	0.5	0.052	0.016	NR	NR	NR	NR	NR
Selenium	5	0.031	0.036	NR	NR	NR	NR	NR
Antimony	10	< 0.003	< 0.003	NR	NR	NR	NR	NR
Mercury	Prohibited	< 0.0002	< 0.0002	< 0.0002	< 0.0002	< 0.0002	< 0.0002	NR
Total Residual Chlorine	Report Max.	NR	NR	NR	NR	NR	NR	0.035
Total Suspended Solids	<=3%	0.58%	0.85%	NR	NR	NR	NR	NR
TSS (NPDES)	50 (max)	NR	NR	NR	NR	NR	NR	10
PH (NPDES)	8.5 (max)	NR	NR	NR	NR	NR	NR	6.37
рН	5.5 - 10.5	6.40	6.2	7.17	7.56	7.35	7.66	NR
Cyanide	0.5	NR	NR	< 0.005	< 0.005	NR	NR	NR
Total Fats, Oil & Grease	<= 300	NR	NR	NR	NR	4.9	18	NR
	1.0 MGD NPDES							
Flow - Gallons/Day	Reg.	66,975	34,500	232	32.5	50	50	0.56 MG
	Any analyte not							
TTO (VOC)	to exceed 1 mg/L	NR	NR	NR	NR	All < 1mg/L	All < 1 mg/L	NR
TTO (ABN)	Any analyte not to exceed 1 mg/L	NR	NR	NR	NR	All < 1mg/L	All < 1mg/L	NR

NR= Not Required

# CWD Monthly Water Quality Analysis

		COMPARISO	ON OF CAMBRI	DGE TAP WATER			
			EPA & STATE				
PARAMETER	Cambridge mg/L	Primary (Health Related) Maximum Contaminant Level(MCL) mg/L	Secondary (Aesthetic Related) MCL mg/L	PARAMETER	Cambridge mg/L	Primary (Health Related) Maximum Contaminant Level(MCL) mg/L	Secondary (Aesthetic Related) MCI mg/L
Alkalinity (as	mg/L	mg/L	mg/L	FAINAIVILILIX	mg/L	mg/L	mg/L
CaCO <sub>3</sub> )	36.5			Magnesium	4.4		
Aluminum	0.018		0.05-0.2	UV254 A/cm	0.025		
Arsenic	<0.018 <0.0005	0.05	0.05-0.2	Manganese	0.025		0.05
Barium	0.0005	2		Mercury	<0.003	0.002	0.05
Darium	0.039			Nitrate (as	<0.0001	0.002	
Cadmium	<0.0005	0.005		Nitrogen)	0.315		
Calcium	21.8	0.003		itili ogen)	0.010		
Chloride	146		250	pH	8.89		
Chlorine, Free	0.01	4.0 MRDL+	200	Selenium	<0.0005	0.05	
Chlorine, Total	2.36	4.0 MRDL+		Silver	<0.00015	0.00	0.1
Chromium	0.001	0.1		Sodium	83		<u> </u>
Color	1		15 color units	Specific Conductance, umhos/cm @25C	606		
Copper	0.002	1.3		Standard Plate Count	0	500 C.F.U./1ml *	250
Dissolved Solids,							
Total	333		500	Sulfate	28		
Fluoride	1.05	4		Total Coliform	0	0 C.F.U./100ml	
Hardness (as CaCO3)	54			Total Haloacetic acids	3.7	60 ug/L (four quarter avg.)	
Iron	0.03		0.3	Total Trihalomethanes	7.8	80 ug/L (four quarter avg.)	
Saturation Index (SI)	-0.16			Turbidity	0.041	0.5 N.T.U filtered, 1.0	5
Lead	0	0.015		Zinc	<0.001		
+ MRDL = Maximu				* C.F.U./1ML = Colo			
Hardness in grains	s per gallon = 3	3		CWD FINISHED WA		COLLECTED:11/	05/09
NA = Not Analyzed				Analyzed by Camb	ridge Lab.		